Reg.No:	Q.P. Code: 16HS604														R16	
SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS) B.Tech I Year II Semester (R16) Regular Examinations May/June 2017 ENGINEERING CHEMISTRY (Common to EEE, ME & CE) (For Students admitted in 2016 only) Time: 3 hours Max. Marks: 60 (Answer all Five Units 5 X 12 = 60 Marks) Image: A write a short note on lead acid batteries. b. Explain construction and working of methanol-oxygen fuel cell. b. Explain the various factors influence the corrosion of metals. b. Explain the various factors influence the corrosion of metals. corr 12M 3 a. Distinguish between hard water and soft water. b. How do you estimate the total hardness of water by EDTA method? corr 10M 4 Describe the Ion-exchange process? LINIT-III 2 a. What are the characteristics of good fuel? b. Describe the characteristics of good fuel? corr 4 b. Describe how synthetic petrol is synthesized from Bergius process. corr 8 b 6 a. Define lubricant. Write the functions of lubricants. b. Write a short note on flash and fire point. cor 6M 7 a. Give the differences between addition and condensation polymerization. b. Write a short note on flash and fire point. cor 6M 8 What are clastomers? Write the preparation and uses of BUNA-S rubber and thikol rubber. cor 12M 9 Define refractoriness. Explain thermal spalling, porosity, dimensional stability and thermal conductivity of the refractories. 12M	Reg.N	lo:														
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